

Spectroscopy - Question Bank - Set 2 (For JAM NET)

* → Very very important in NET

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- * ① Show that the transition $A_1 \rightarrow A_2$ is forbidden for electric dipole transition in NH_3 molecule.
- ② Show that the function xy has B_2 symmetry properties in group C_{3v} .
- ③ Does the integral $\int (3d_{z^2})x (3d_{xy}) d\tau$ vanish in a T_d symmetry molecule?
- * ④ Is $p_y \rightarrow p_x$ an allowed transition in a T_d molecule?
- ⑤ What are the allowed transitions and their polarization of a B_1 electron in a C_{3v} system?
- * ⑥ Investigate whether an A_1 electron in H_2O can make an electric dipole transition to a B_1 orbital.
- * ⑦ The ClO_2 molecule was trapped in a solid. Its ground state is known to be B_1 . Light polarized parallel to the y -axis excited the molecule to an upper state. What is the symmetry of the state?
- * ⑧ The ground state of NO_2 is A_1 . To what excited state may it be excited by electric dipole transition and what polarization of light is necessary to use?
- * ⑨ Check the $\delta - \delta^*$ transition is allowed or not for the following molecule.

$$\left[\text{Re}_2\text{Cl}_8 \right]^{2-} \quad \left[\text{Mo}_2\text{Cl}_8 \right]^{4-}$$
- ⑩ What are the selection rules for electric dipole transition and magnetic dipole transition. Explain with proper examples.

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- * (11) One photon transition and two photon transition - Explain the selection rules by proper diagram.
- * (12) What is double resonance? How to calculate the irradiation frequency for a chosen nuclear system?
- * (13) What do you mean by "x polarized / y polarized or (x,y) polarized"?
- * (14) What is totally symmetric modes? Explain its significance in spectroscopic transition.
- (15) What is Σ , Δ , Π ?
- * (16) $\sum \rightarrow \sum$ type contour: Explain transition produce PR
- (17) Please check the Correct transition (allotted). Explain with proper reason.

