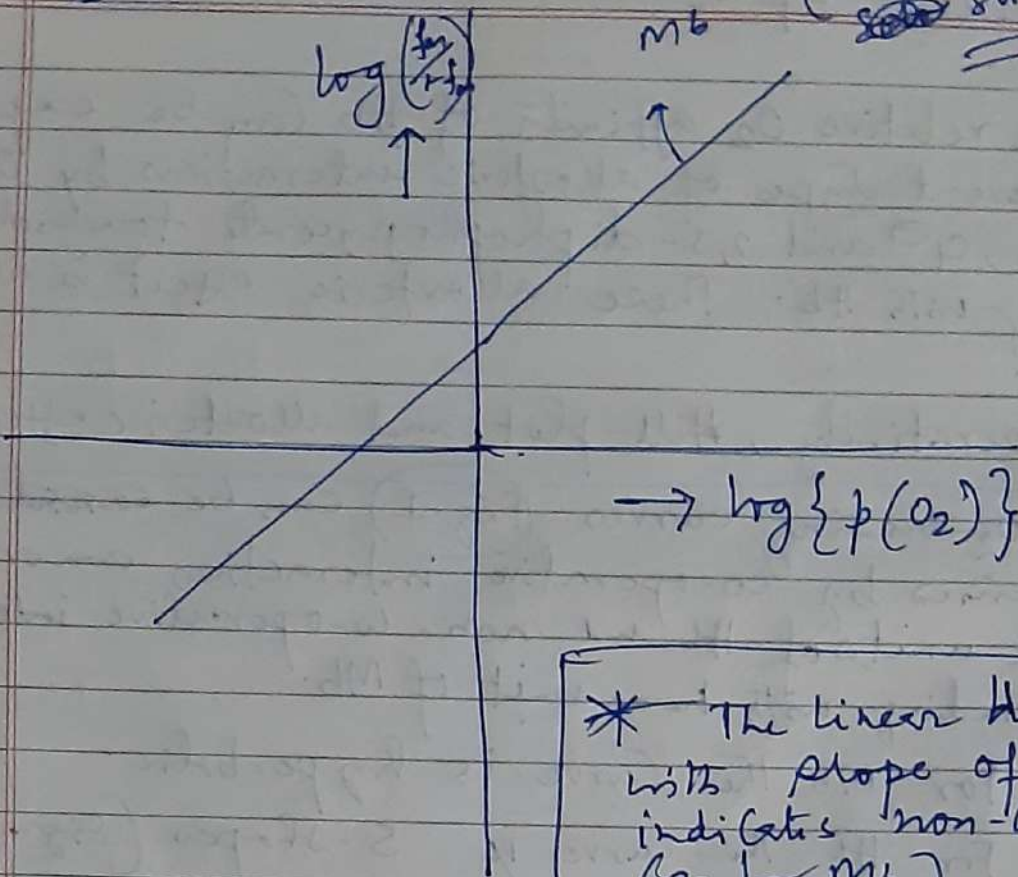


For Mb

Hill plot for Mb

(Non-cooperative)
~~sub~~ steepest

(Homotropic
allosteric
interactions are
absent in Mb.)



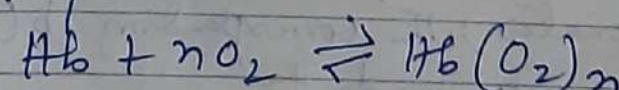
* The linear Hill plot with slope of unity indicates non-cooperativity (as for Mb)

$$\log \left(\frac{f_m}{1-f_m} \right) = \log \{ p(O_2) \} - \log p_{50}$$

(slope = 1)

For Hb :-

In case of Hb, the corresponding expressions are complicated and results are empirically formulated as follows for Hb protein



$$K_H = \frac{[Hb(O_2)_n]}{[Hb] \{ p(O_2) \}^n} = \frac{f_H}{(1-f_H) \{ p(O_2) \}^n} = \frac{1}{(p_{50})^n}$$

$$\text{Then, } f_H = \frac{K_H \{ p(O_2) \}^n}{[1 + K_H \{ p(O_2) \}^n]}$$

$$\therefore \log \frac{f_H}{1-f_H} = n \log \{ p(O_2) \} + \log K_H = n \log \{ p(O_2) \} - n \log(p_{50})$$

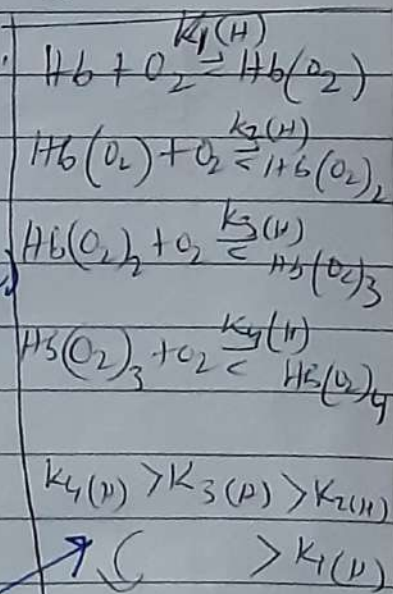
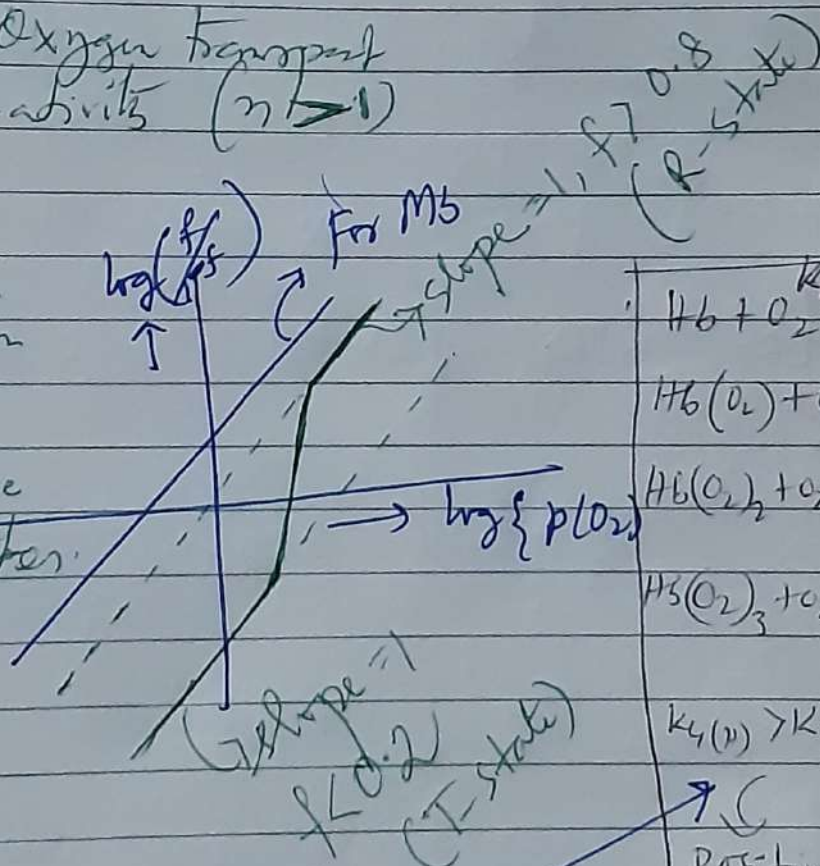
For Hb, $n > 1$, and it indicates that O_2 -binding in the subunits of Hb is ~~interdependent~~ interdependent and it suggests positive cooperativity among the heme units due to heme-heme interaction.

To characterize the co-operativity for Hb, the O_2 binding data at intermediate saturations (i.e., $0.2 < f < 0.8$) yield a straight line in Hill plot with a slope close to 3.

$f < 0.2$ & $f > 0.8$ (No cooperativity)
i.e., $n=1$

* Generally, for Oxygen transport protein cooperativity ($n > 1$)
Hb, Hc

For oxygen storage protein (Mb, Hb) non cooperative effect predominates ($n=1$).



Homotropic allosteric interaction

positive cooperative effect

* Will be Discussed in next lecture \Rightarrow

- Home tasks:
- ① What is T State & R-State?
 - ② What is α -chain & β -chain?
 - ③ What is triggered mechanism?
 - ④ What is Soret band & α, β band? What are their position in UV-Vis spectrum?
 - ⑤ Magnetic properties and spin states in deoxy Hb & oxy Hb.

3. Elementary quantum mechanics

Q-1 :- Define Black-body radiation and explain its characteristics features.

Q-2 :- Explain the Planck's quantum theory.

Q-3 :- Define photoelectric effect and explain the law of photoelectric emissions.

Q-4 :- Derive Einstein photo emission equation.

Q-5 :- Explain the de-Broglie's hypothesis.

Q-6 :- Derive time independent Schrödinger equation.

Q-7 :- State and explain Heisenberg's uncertainty principle.

Q-8 :- Derive time dependent Schrödinger equation.

Q-9 :- Give the physical significance of $\Psi(x, y, z, t)$ and $\Psi^2(x, y, z, t)$.

Q-10 :- What is normalized wave function.

Q-11 :- State the basic postulates of quantum mechanics.

Q-12 :- What is an operator?

Q-13 :- Define linear operator?

Q-14 :- Enlist the rules for setting up quantum mechanical operators.

Q-15 :- What is Commutator of the operators?

Q-16 :- Evaluate the following Commutators.

$$[\hat{x}, \frac{d}{dx}], [\hat{x}, \hat{p}_x], [\frac{\partial}{\partial x}, \frac{\partial}{\partial y}], [\hat{x}, \hat{H}_x], [\hat{p}_x, \hat{H}_x], [\hat{p}_x^2, \hat{x}^2],$$

Q-17 :- What is Eigen value equation?

Q-18 :- Which of the following functions

$$\sin 3x$$

$$\ln 2x$$

$$x \sin x$$

$$3e^{-5x}$$

are eigen functions of the operator $\frac{\partial^2}{\partial x^2}$.

Q-19:- Show that e^{+ikx} and e^{-ikx} are eigen function of one dimensional linear momentum operator and what are the corresponding eigen values?

Q-20:- What is stationary states?

Q-21:- What are the characteristics that must be satisfied by a function to be an eigen?

Q-22:- Indicate which of the following functions are acceptable wave function:

- i) $\psi = x$, ii) $\psi = x^2$, iii) $\psi = e^x$, iv) $\psi = e^{-x}$, v) $\psi = e^{-x^2}$
vi) $\sin^{-1}x$ (-1, 1)

Q-23:- Prove that if $\hat{\alpha}$ and $\hat{\beta}$ are two linear operators then $(\hat{\alpha} + \hat{\beta})$ and $\hat{\alpha}\hat{\beta}$ are also linear operators.

Q-24:- Which of the following wave functions are acceptable in quantum mechanics?

- i) $\sin x$, ii) $\tan x$, iii) $\operatorname{cosec} x$, iv) $\sin x + \cos x$

Q-25:- If ψ_1 and ψ_2 are eigen functions of linear operator \hat{A} with same eigen values a , show that any linear combination of ψ_1 and ψ_2 will also be an eigen function of \hat{A} with same eigen value.

Q-26:- If $\psi_1(x, t)$ and $\psi_2(x, t)$ are both solutions of time dependent Schrodinger wave equation for a given potential $V(x)$. Show that $\psi(x, t) = a_1\psi_1 + a_2\psi_2$ is also a solution, where a_1 and a_2 are constants.

Q-27:- Define Hermitian Operator? What is the condition for an operator to be Hermitian?

Q-28:- Check the following operators whether Hermitian or Anti Hermitian in nature.

- i) $\frac{d}{dx}$, ii) $\frac{d^2}{dx^2}$, iii) \hat{p}_x , iv) \hat{H}_x , v) $\hat{\alpha}\hat{\beta} - \hat{\beta}\hat{\alpha}$, vi) $i(\hat{\alpha}\hat{\beta} - \hat{\beta}\hat{\alpha})$
- } $\hat{\alpha}, \hat{\beta}$
} Hermitian operator

Q-29:- If two operators $\hat{\alpha}, \hat{\beta}$ are Hermitian, then find out its conditions, when $\hat{\alpha}\hat{\beta}$ will be Hermitian.

Q-30:- What is orthonormal wave function?

Q-31:- Prove that eigen ~~function~~ values of Hermitian operators are real.

Q-32:- If $\hat{\alpha}$ & $\hat{\beta}$ are a pair of Commuting Hermitian operators, then they possess a common set of eigen functions.

Q-33:- If two Hermitian operators \hat{A} & \hat{B} possess a common set of eigen functions ψ_i - then they commute.

Q-34:- If $\psi_m(x)$ & $\psi_n(x)$ are non-degenerate eigen functions of a Hermitian operator then they are orthogonal.

Q-35:- Write down the Schrödinger wave equation of a free particle moving in one dimensional box and find out the wave functions and also calculate the allowed energy.

Q-36:- Find out the normalised wave function for free particle moving in one dimensional potential box.

Q-37:- Show that the free particle wavefunctions are orthogonal with each other.

Q-38:- Draw the wave function of free particle for $n=1, 2, 3$.

Q-39:- Draw the probability function of free particle for $n=1, 2, 3$.

Q-40:- Is the energy of a free particle quantised? Justify.

Q-41:- What is zero point energy of free particle?

Q-42:- Show that the energy of the free particle is in the agreement with the uncertainty principle.

✓ Q-43:- Discuss the motion of a free particle in the three dimensional potential box.

Q-44:- For a particle in a cubical box, write down the energy value for the condition, $n_x + n_y + n_z = 4$ and indicate the level of degeneracy if any.

✓ Q-45:- Evaluate the expectation value of K.E of a free particle in an one dimensional potential box of length 'l'.

✓ v.v.f Q-46:- Calculate the expectation value of $\langle \hat{x} \rangle$, $\langle \hat{x}^2 \rangle$, $\langle \hat{p}_x \rangle$ and $\langle \hat{p}_x^2 \rangle$ for a particle in a one dimensional box of width 'l' and infinitely high potential walls with origin at the end of the box.

✓ Q-47:- Derive the energy expression of Simple Harmonic Oscillator (SHO).

✓ Q-48:- Show that the ~~energy~~ zero point energy of the ~~simple~~ harmonic oscillator is consistent with the uncertainty principle.

* Q-49:- Verify that the function $A \cdot \exp(-\frac{1}{2} \beta x^2)$ is an eigen-function of a linear harmonic oscillator, where $\beta = (mk)^{1/2} / \hbar$

Q-50:- Write down the Schrodinger time independent equation of wave for H atom in terms of both Cartesian & polar coordinates.

* Q-51:- Show that energy of H-atom is n^2 -fold degenerate.

Q-52:- Show that H-atom wave functions are orthogonal to each other.

Q-53:- What is sinusoidal wave function?

Q-54:- From the probability density of $1s e^-$ show that the most probable distance of e^- from the nucleus is equal to Bohr radius.

$$\Psi_{1s} = \frac{1}{(\pi a_0^3)^{1/2}} \cdot e^{-r/a_0}, \text{ where } a_0 \text{ is the Bohr radius.}$$

Q-55:- For the $2s$ state of H atom,

$$\Psi_{2s} = \frac{1}{\sqrt{32\pi}} (2-r) \cdot e^{-r/2} \text{ in a.u., find the most probable and the nodal distances of the electron from the nucleus.}$$

Q-56:- Normalise the ground state wave function for one dimensional harmonic oscillator $\Psi_0 = f(x) = A \cdot e^{-\frac{\beta x^2}{2}}$

$$\text{Where } \beta = \frac{(mk)^{1/2}}{\hbar}$$

Q-57:- Show that the following two wave functions are orthogonal to each other. $\Psi_0(x) = \left(\frac{\beta}{\pi}\right)^{1/4} e^{-\frac{\beta x^2}{2}}$ and $\Psi_1(x) = \left(\frac{4\beta^3}{\pi}\right)^{1/4} x \cdot e^{-\frac{\beta x^2}{2}}$, β is a constant.

Q-58:- Derive the bonding & antibonding ^{orbital} wavefunction for sp , sp^2 & sp^3 orbital.

Q-59:- ~~test~~ Define radial wave function & angular wave function.

Q-60:- What is Linear Combination of Atomic Orbital. Explain with example.